

NOMENCLATURE REVIEW

NAME THE FOLLOWING COMPOUNDS:

1. BaSO_3
2. $(\text{NH}_4)_3\text{PO}_4$
3. PBr_5
4. MgSO_4
5. CaO
6. H_3PO_4
7. $\text{Na}_2\text{Cr}_2\text{O}_7$
8. MgO
9. SO_3
10. $\text{Cu}(\text{NO}_3)_2$
11. HI
12. N_2O
13. MnO
14. AgNO_3
15. As_2O_5
16. Fe_2O_3
17. HClO
18. N_2O_3
19. HF
20. $\text{H}_2\text{C}_2\text{O}_4$
21. NaHCO_3
22. SiBr_4
23. CuCl_2
24. HNO_2
25. SnO_2
26. BaCrO_4

WRITE FORMULAS FOR THE FOLLOWING COMPOUNDS:

27. hydrobromic acid
28. chromium(III) carbonate
29. magnesium sulfide
30. iodine trichloride
31. lithium hydride
32. ammonium hydroxide
33. calcium chloride
34. hydroselenic acid
35. iron(II) nitride
36. aluminum hydroxide
37. tin(II) fluoride
38. sulfur tetrachloride
39. mercury(II) iodide
40. diphosphorus pentoxide
41. sulfurous acid
42. lead(II) nitrate
43. dihydrogen monoxide
44. sodium oxalate
45. perchloric acid
46. chlorous acid
47. silicon dioxide
48. carbonic acid
49. sodium chlorate
50. xenon hexafluoride
51. nickel nitrate
52. potassium perchlorate
53. tin(IV) carbonite pentahydrate
54. iron(III) hydrogen perphosphate
55. $\text{Cr}_2(\text{CO})_3 \cdot 3\text{H}_2\text{O}$
56. MnHPO_3
57. MgCrO_4

NOMENCLATURE REVIEW

ANSWER KEY

1. barium sulfite
2. ammonium phosphate
3. phosphorus pentabromide
4. magnesium sulfate
5. calcium oxide
6. phosphoric acid
7. sodium dichromate
8. magnesium oxide
9. sulfur trioxide
10. copper(II) nitrate
11. hydroiodic acid
12. dinitrogen monoxide
13. manganese(II) oxide
14. silver nitrate
15. diarsenic pentoxide
16. iron(III) oxide
17. hypochlorous acid
18. dinitrogen trioxide
19. hydrofluoric acid
20. oxalic acid
21. sodium bicarbonate
22. silicon tetrabromide
23. copper(II) chloride
24. nitrous acid
25. tin(IV) oxide
26. barium chromate
27. HBr
28. $\text{Cr}_2(\text{CO}_3)_3$
29. MgS
30. ICl_3
31. LiH
32. NH_4OH
33. CaCl_2
34. H_2Se
35. Fe_3N_2
36. $\text{Al}(\text{OH})_3$
37. SnF_2
38. SCl_4
39. HgI_2
40. P_2O_5
41. H_2SO_3
42. $\text{Pb}(\text{NO}_3)_2$
43. H_2O
44. $\text{Na}_2\text{C}_2\text{O}_4$
45. HClO_4
46. HClO_2
47. SiO_2
48. H_2CO_3
49. NaClO_3
50. XeF_6
51. $\text{Ni}(\text{NO}_3)_2$
52. KClO_4
53. $\text{Sn}(\text{CO}_2)_2 \cdot 5\text{H}_2\text{O}$
54. Li_2O_2
55. $\text{Fe}_2(\text{HPO}_4)_3$
56. Chromium (III) hypocarbonite trihydrate
57. Manganese (II) hydrogen phosphite
58. Magnesium chromate