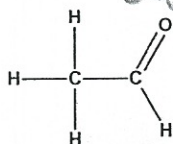


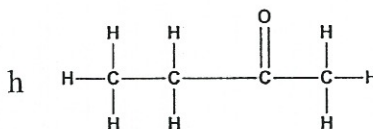
Organic Compounds and Functional groups

1. Name the type of organic compound each of the following would be classified as:

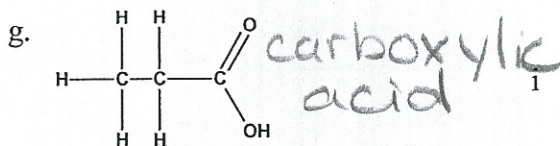
- a. Propanol *alcohol*
- b. Aminoethane *amine*
- c. Butanone *ketone*
- d. Methoxymethane *ether*
- e. Ethanal *aldehyde*
- f.



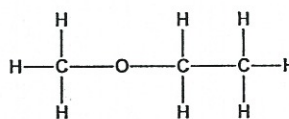
aldehyde



ketone

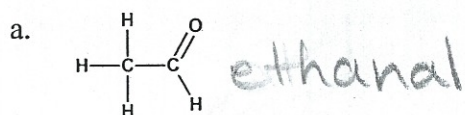


carboxylic acid

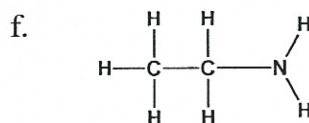


ether

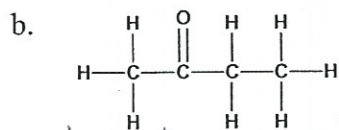
2. Write the name for each of the following:



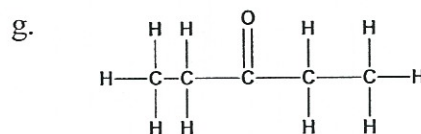
ethanal



ethanamine



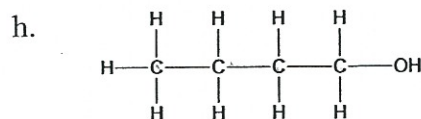
butanone



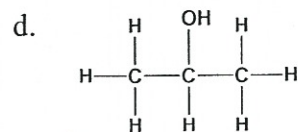
3-pentanone



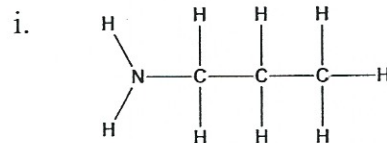
methanal



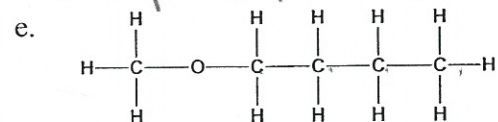
1-butanol



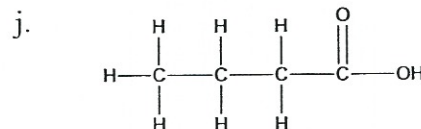
2-propanol



propanamine

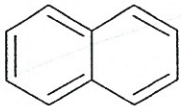



1-methoxybutane



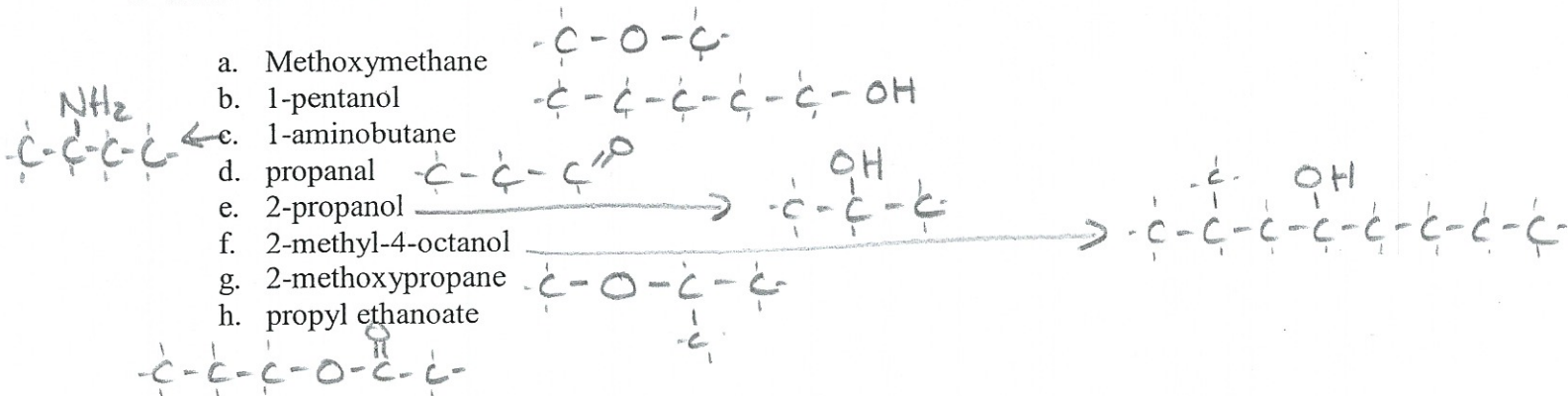
butanoic acid

3. Name the type of organic compound each of the following would be classified as:

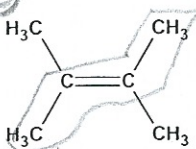
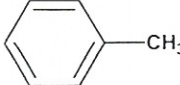

- a.  aromatic
- b. $\text{H}_3\text{C}-\text{CH}_2-\text{C}(=\text{O})-\text{O}-\text{CH}_2-\text{CH}_3$ ester (ethyl propanoate)
- c. $\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{OH}$ alcohol (1-propanol)
- d. $\text{H}_3\text{C}-\text{C}\equiv\text{C}-\text{CH}_3$ alkyne
- e. $\text{C}_{16}\text{H}_{34}$ alkane
- f. $\text{H}_3\text{C}-\text{CH}_2-\text{O}-\text{CH}_2-\text{CH}_3$ ether (ethoxyethane)
- g. $\text{H}_3\text{C}-\text{CH}_2-\text{C}(=\text{O})-\text{H}$ aldehyde (propanal)
- h. $\text{H}_3\text{C}-\text{C}(=\text{O})-\text{CH}_3$ ketone (propanone)
- i.  cyclic hydrocarbon (cyclopentane)
- j. $\text{H}-\text{C}(=\text{O})-\text{OH}$ carboxylic acid (methanoic acid)
- k. $\text{H}-\text{C}(=\text{O})-\text{OH}$ "
- l. $\text{H}_2\text{C}=\text{CH}-\text{CH}_2-\text{CH}_3$ alkene (1-butene)

4. Draw the structural formula for:

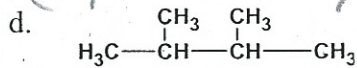
- a. Methoxymethane
- b. 1-pentanol
- c. 1-aminobutane
- d. propanal
- e. 2-propanol
- f. 2-methyl-4-octanol
- g. 2-methoxypropane
- h. propyl ethanoate



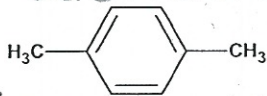
5. Name the type of organic compound each of the following would be classified as:

- a. $\text{H}_3\text{C}-\text{C}(=\text{O})-\text{CH}_2-\text{CH}_2-\text{CH}_3$ ketone (2-pentanone)
- b. $\text{H}_3\text{C}-\text{CH}_2-\text{CH}=\text{C}(\text{CH}_3)-\text{CH}_3$ alkene (2-methyl-2-pentene)
- c. $\text{HO}-\text{C}(=\text{O})-\text{CH}_2-\text{CH}_3$ carboxylic acid (propanoic acid)
- k.  alkene (2,3-dimethyl-2-butene)
- l.  aromatic (methyl benzene)
- m.  cyclic hydrocarbon (cyclopentene)

alkane
(2,3-dimethylbutane)

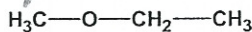


e. aromatic

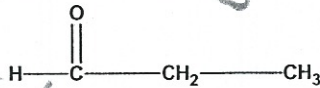


(1,4-dimethyl benzene
or p-dimethyl benzene)

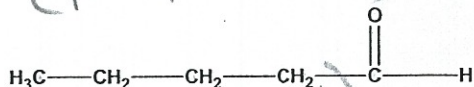
f. ether
(methoxy ethane)



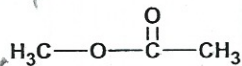
g. aldehyde
(propanal)



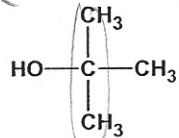
h. aldehyde
(pentanal)



i. ester
(methyl ethanoate)



j. alcohol
(methyl-2-propanol)

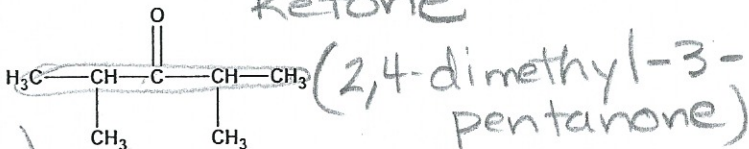


class (name)

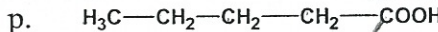
cyclic hydrocarbon
(cyclopentane)



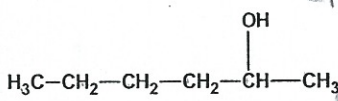
ketone



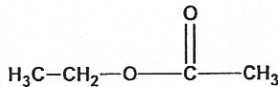
carboxylic acid
(pentanoic acid)



q. alcohol
(2-hexanol)



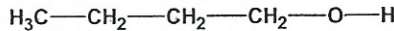
r. ester
(ethyl ethanoate)



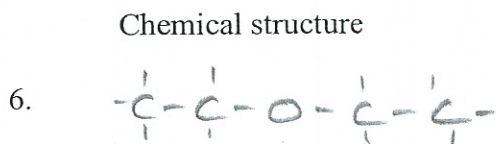
s. alcohol
(ethanol)



t. alcohol
(1-butanol)

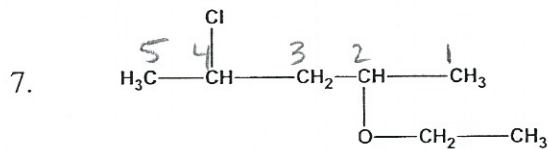


Name or draw the following compounds:

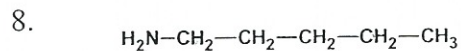


IUPAC name

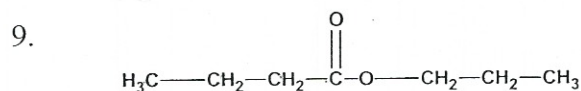
Ethoxyethane



2-ethoxy 4-chloro pentane



1-pentanamine



propyl butanoate