Carboxylic Acids and Esters



Carboxyl Group

Carboxylic acids contain the carboxyl group on carbon 1. O \parallel $CH_3 - C - OH = CH_3 - COOH$ carboxyl group

Naming Carboxylic Acids

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IUPAC alkan -oic acid

Common prefix – ic acid

HCOOH CH₃COOH CH₃CH₂COOH CH₃CH₂COOH methanoic acid ethanoic acid propanoic acid

butanoic acid

formic acid acetic acid propionic acid butyric acid

Naming Rules

- Identify longest chain
- (IUPAC) Number carboxyl carbon as 1

 $\begin{array}{l} \mathsf{CH}_3\\ \mathsf{I}\\ \mathsf{CH}_3 - \mathsf{CH} - \mathsf{CH}_2 - \mathsf{COOH}\\ \end{array}$

Learning Check

Give IUPAC names:

A. CH₃COOH

CH₃

B. CH₃CHCOOH

Solution

 A. CH₃COOH ethanoic acid (acetic acid) CH₃ |
 B. CH₃CHCOOH 2-methylpropanoic acid

Esters

In an ester, the H in the carboxyl group is replaced with an alkyl group



Naming Esters

- Name the alkyl from the alcohol –O-
- Name the acid with the C=O with –oate acid alcohol
 O |

$$H = 1$$
 methyl
 $CH_3 - C = 0$ - CH_3
ethanoate H_3 methyl

(acetate)

methyl ethanoate (IUPAC) methyl acetate (common)

Some Esters and Their Names

Flavor/Odor Raspberries HCOOCH₂CH₃

ethyl methanoate (IUPAC) ethyl formate (common)

Pineapples CH₃CH₂CH₂COOCH₂CH₃ ethyl butanoate (IUPAC) ethyl butyrate (common)

Learning Check

Give the IUPAC names of the following compound, which is responsible for the flavor and odour of pears.



Solution

$\begin{array}{c|c} & & & propyl \\ \hline & & & Propyl \\ \hline & & & C & - C & - C & - C & + 2 &$

Learning Check

Draw the structure of the following compounds:

A. 3-bromobutanoic acid

B. Ethyl propanoate

Solution

A. 3-bromobutanoic acid Br I CH₃CHCH₂COOH

B. Ethyl propanoate O I CH₃ CH₂ COCH₂CH₃ CH₃CH₂COOCH₂CH₃