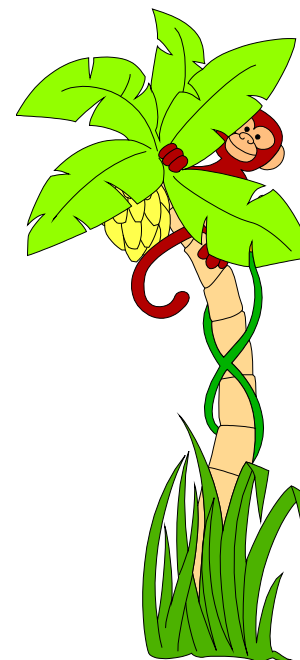


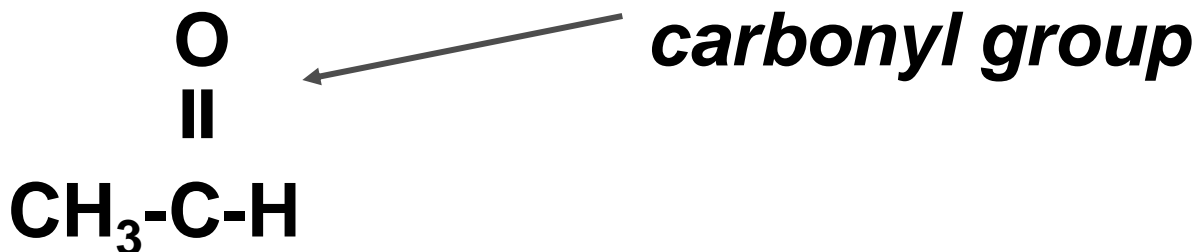
Aldehydes and Ketones

Naming, Physical Properties and Reactions

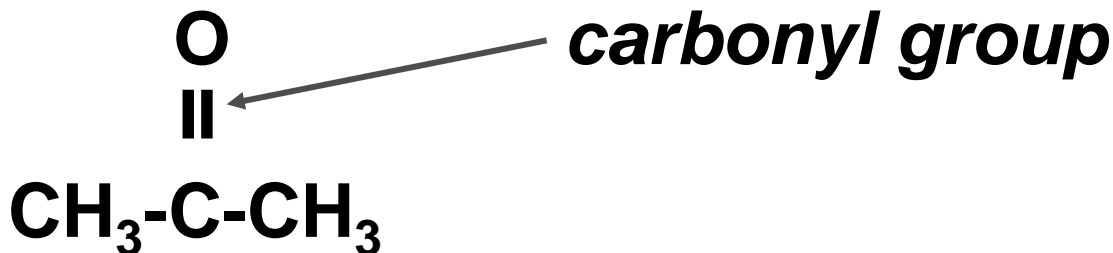


Aldehydes and Ketones

- In an aldehyde, the double bonded oxygen is found on the end of the C chain

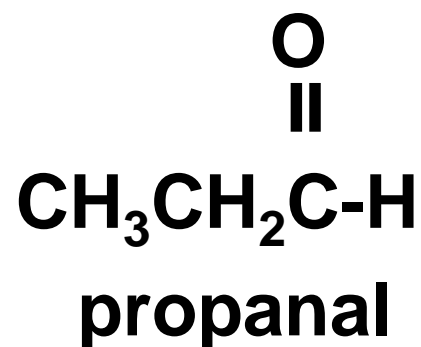
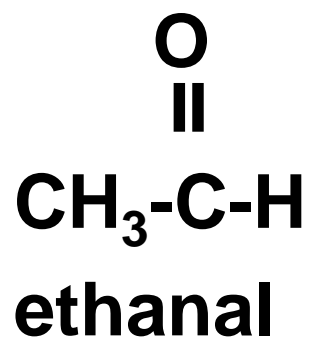
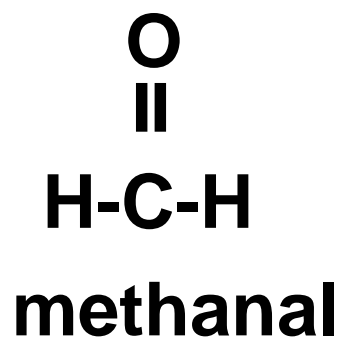


- In a ketone, the double bonded oxygen is found in the middle of the C chain



Naming Aldehydes

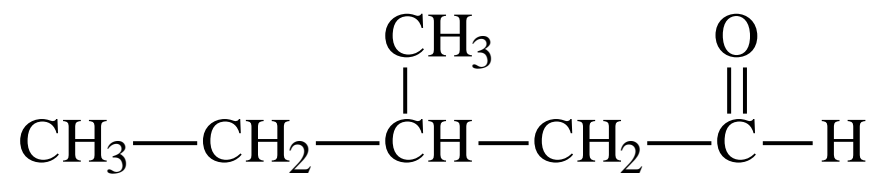
- IUPAC Replace the -e in the alkane name with **-al**



Common names:

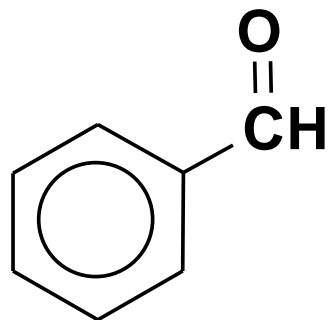
(formaldehyde) (acetaldehyde) (propionaldehyde)

Example

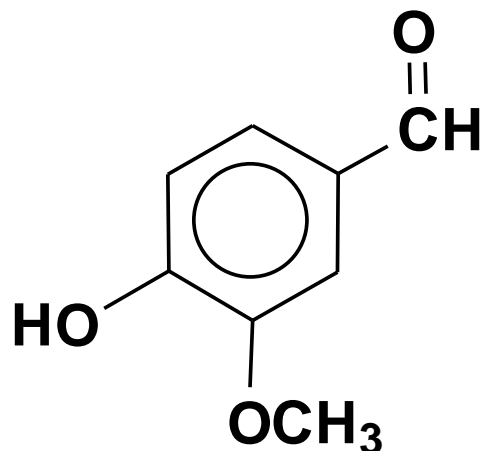


3-methylpentanal

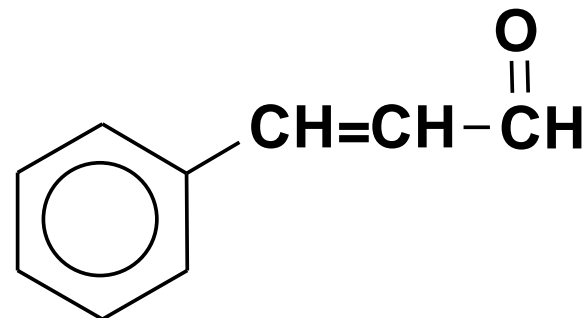
Aldehydes as Flavorings



Benzaldehyde
(almonds)

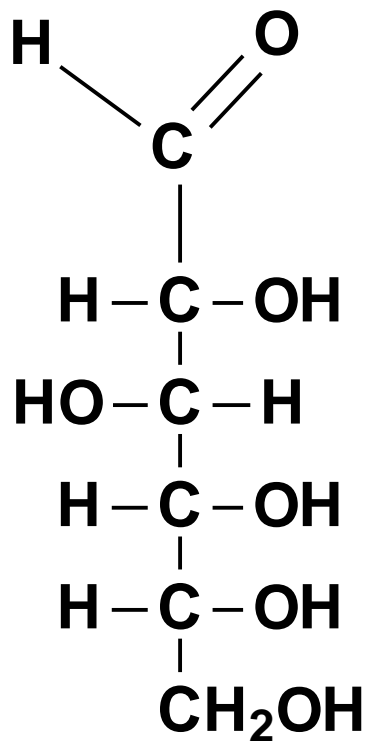


Vanillin
(vanilla beans)



Cinnamaldehyde
(cinnamon)

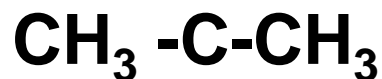
Glucose is an aldehyde



glucose

Naming Ketones

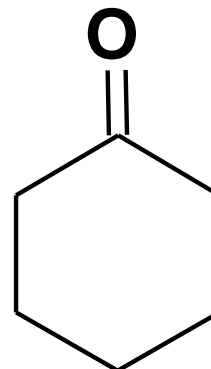
- In the IUPAC name, the -e in the alkane name is replaced with **-one**



Propanone



2-Butanone

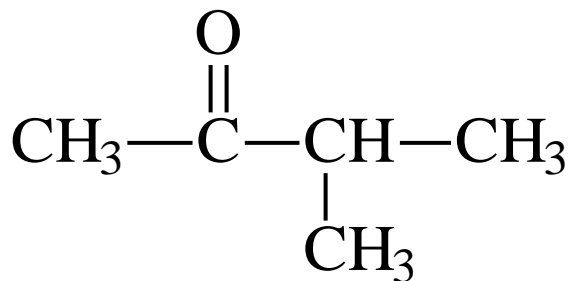


Cyclohexanone

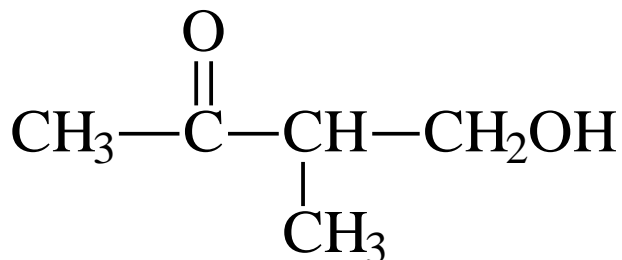
Common names:

(Dimethyl ketone) (Ethyl methyl ketone)

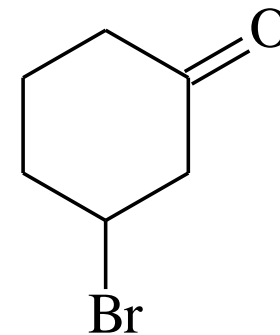
Examples



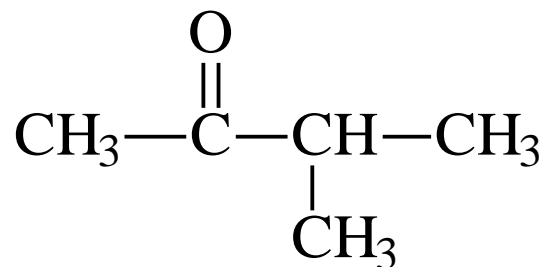
3-methyl-2-butanone



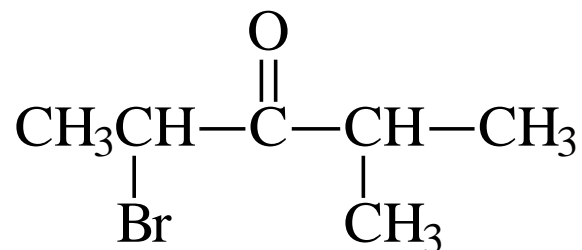
4-hydroxy-3-methyl-2-butanone



3-bromocyclohexanone

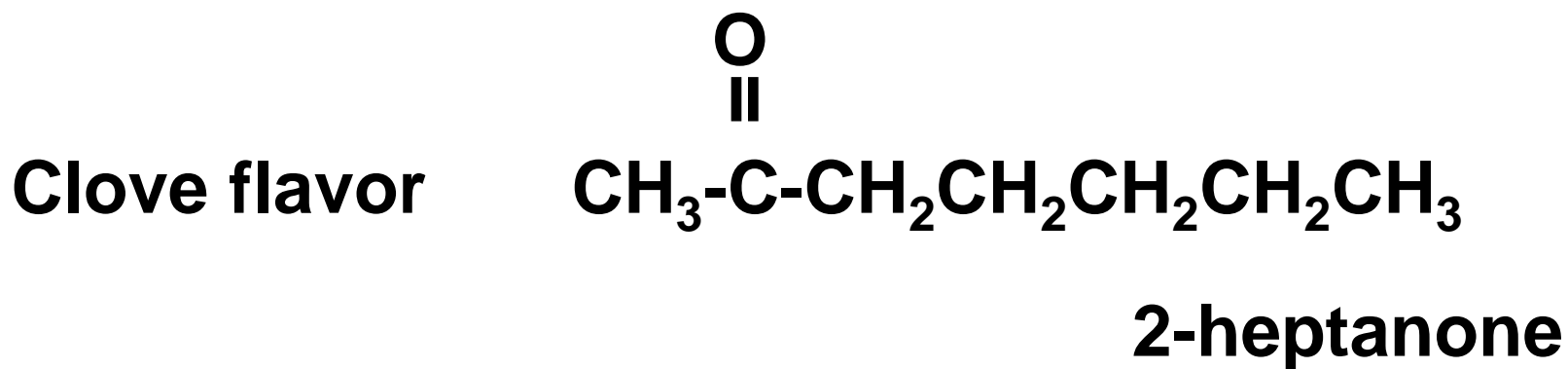
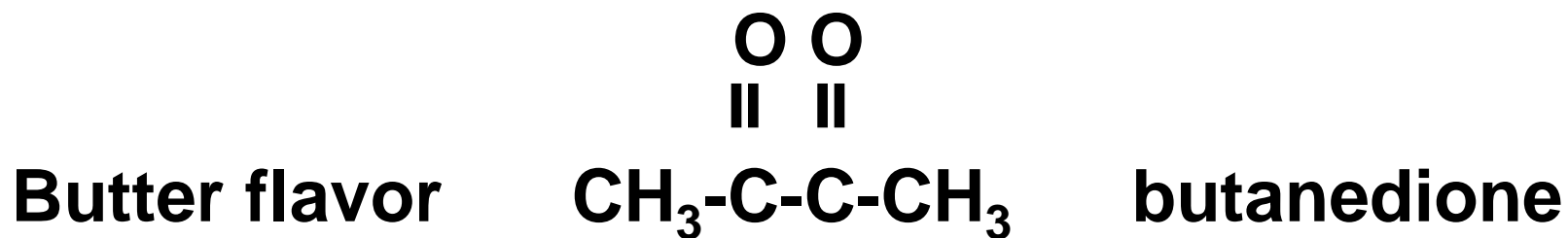


3-methyl-2-butanone

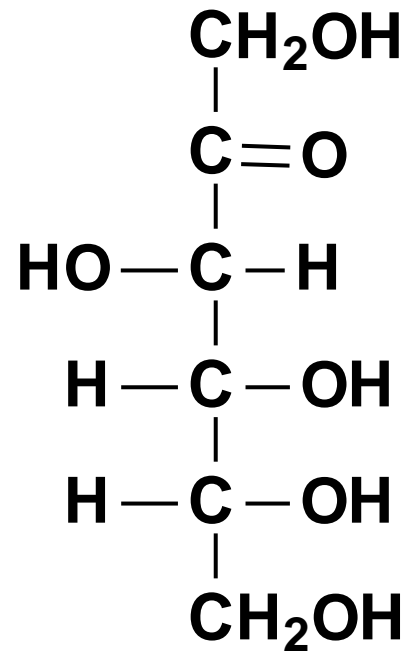


2-bromo-4-methyl-3-pentanone

Ketones

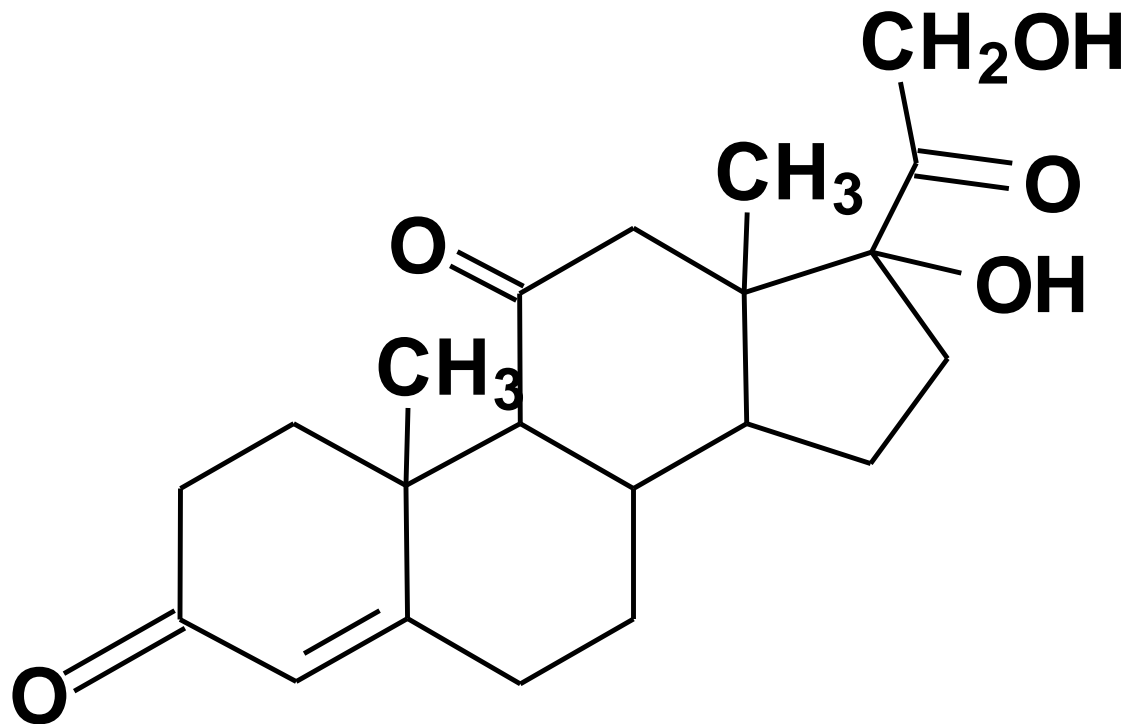


Fructose is a Ketone



D-Fructose

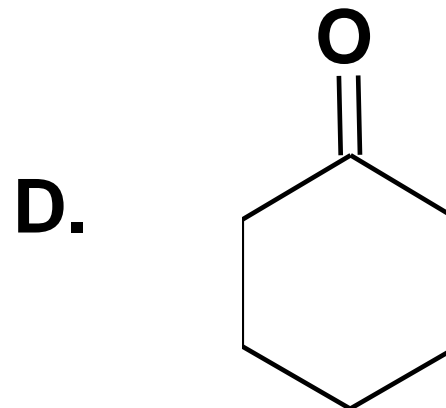
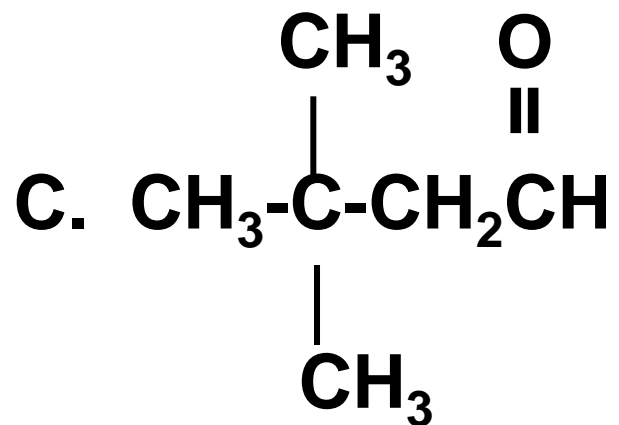
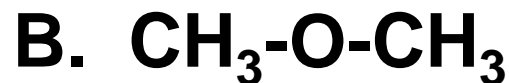
Ketones as Hormones



Cortisone

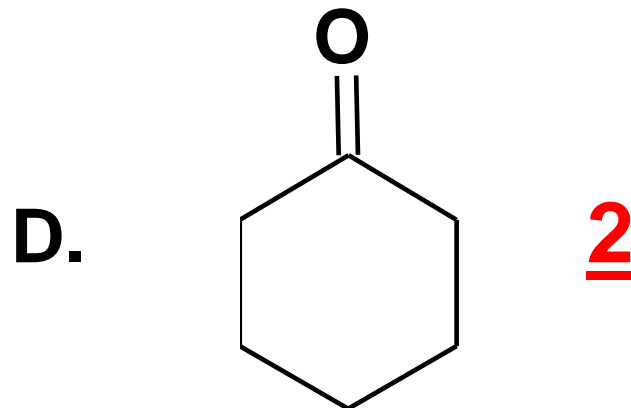
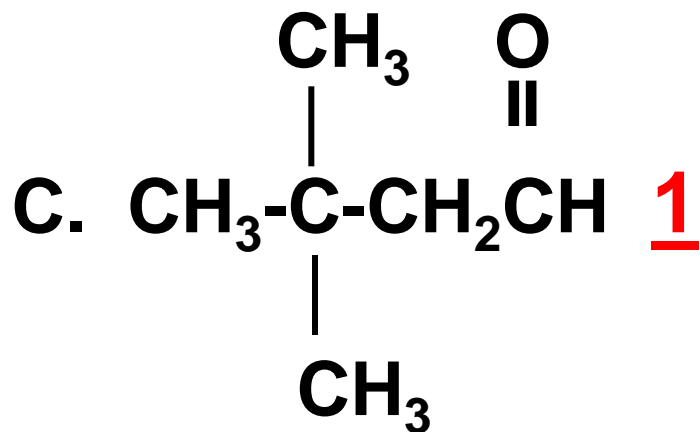
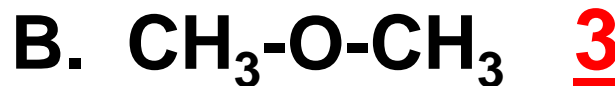
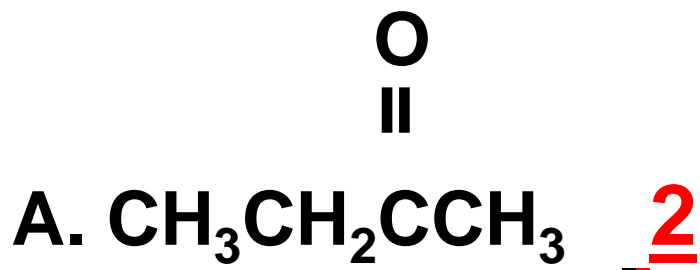
Learning Check

Classify each as an aldehyde (1), ketone (2) or neither(3).



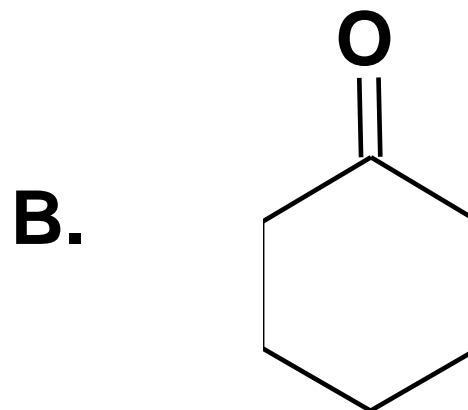
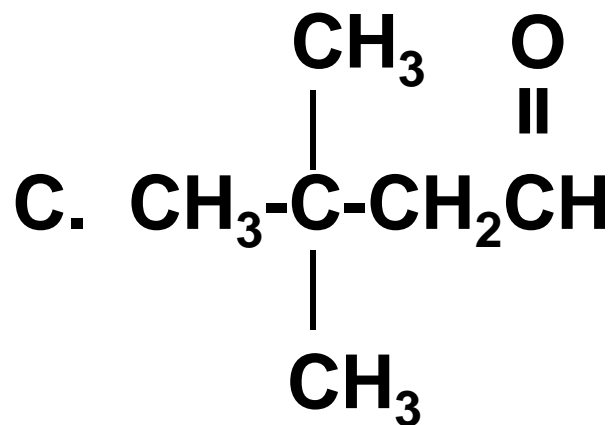
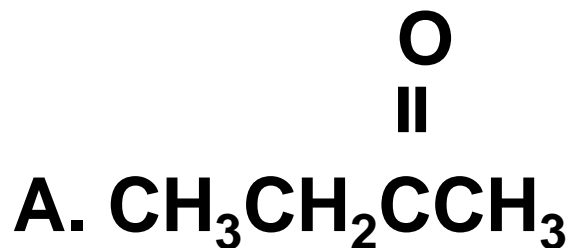
Solution

Classify each as an aldehyde (1), ketone (2) or neither(3).

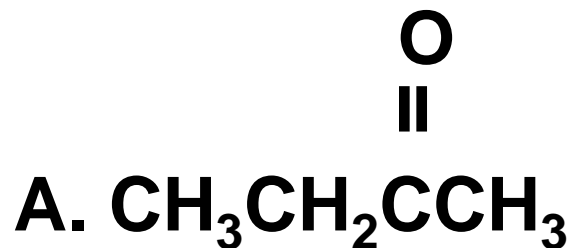


Learning Check

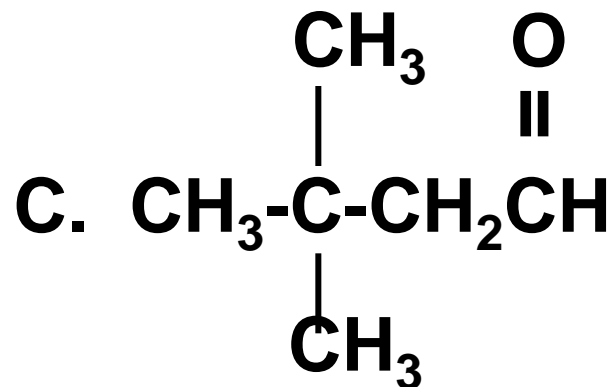
Name the following



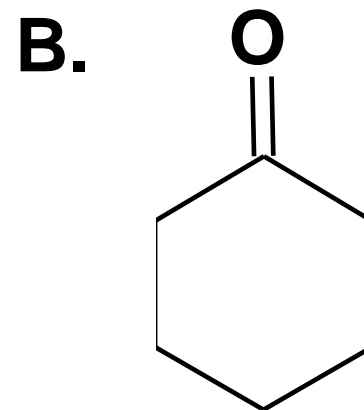
Solution



2-butanone (ethyl methyl ketone)



3,3-dimethylbutanal



cyclohexanone

Learning Check

Draw the structural formulas for each:

A. 4-methylpentanal

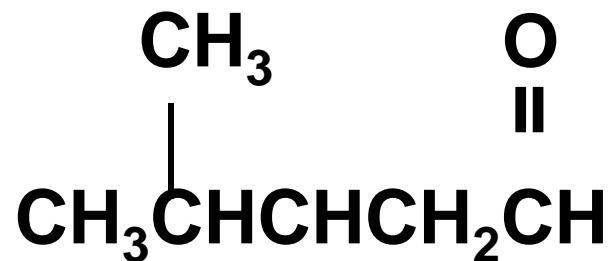
B. 2,3-dibromopropanal

C. 3-methyl-2-butanone

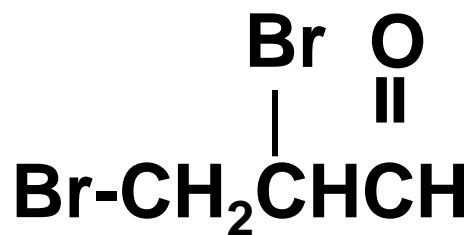
Solution

Draw the structural formulas for each:

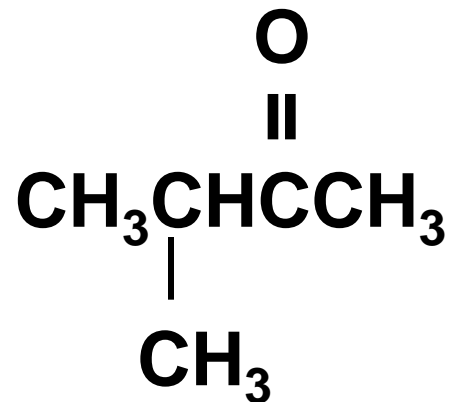
A. 4-methylpentanal



B. 2,3-dibromopropanal



C. 3-methyl-2-butanone



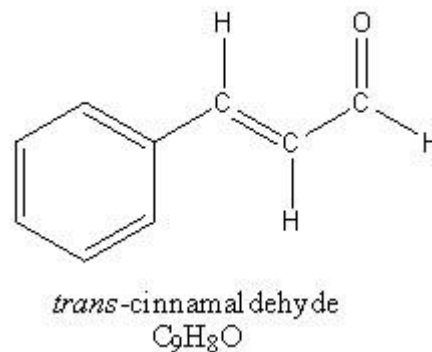
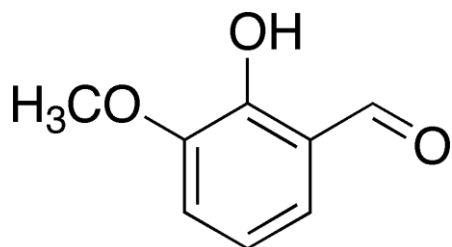
Properties of Ketones and Aldehydes

- Boiling and melting points are between alkanes and alcohols
- Very reactive
- Slightly polar
- Distinctive odours
- Only short-chain ones are soluble in water

Uses of Ketones and Aldehydes

- Solvents ex. Propanone (acetone)
- Flavourings ex. Vanillin, cinnamaldehyde, almond flavour

vanillin



- Used in the manufacture of plastics and adhesives
- Preservative for biological specimens ex. Methanal or formaldehyde